

2-Bromo-3,4,5-trimethoxybenzyl 4-bromo-7-methoxy-1,3-benzodioxole-5-carboxylate

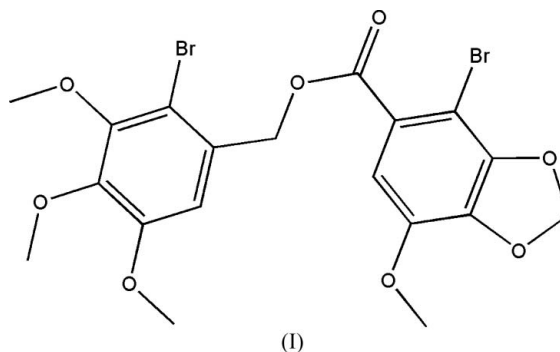
Xian-Shu Fu,^{a*} Yan Zhang,^b
Jun-Wei Sun^a and Xiao-Xia Yu^a^aCollege of Life Sciences, China Jiliang University, Hangzhou 310018, People's Republic of China, and ^bCollege of Foreign Languages, China Jiliang University, Hangzhou 310018, People's Republic of ChinaCorrespondence e-mail:
fuxianshu2003@yahoo.com.cn

Key indicators

Single-crystal X-ray study
 $T = 294$ K
Mean $\sigma(\text{C}-\text{C}) = 0.004$ Å
 R factor = 0.032
 wR factor = 0.078
Data-to-parameter ratio = 17.4For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.The title compound, $\text{C}_{19}\text{H}_{18}\text{Br}_2\text{O}_8$, was synthesized in an anhydrous medium at 293 K. The five-membered ring adopts an envelope conformation. The dihedral angle between the two benzene rings is $2.6 (1)^\circ$.Received 15 June 2006
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Comment

The title compound, (I), is a key intermediate of biphenyl derivatives, which may act to moderate liver ailments and thus be effective in the treatment of acute and chronic hepatitis (Song & Xiao, 1982).



The molecular structure of (I) is illustrated in Fig. 1. The bond lengths and angles are unremarkable. The five-membered ring adopts an envelope conformation with atom C7 as the flap. The dihedral angle between the C1–C6/O2/O3 and C11–C16 planes is $2.6 (1)^\circ$, and the angle between the C9/C10/O4/O5 and C1–C6/O2/O3 planes is $21.8 (2)^\circ$. The C–O–C–C torsion angles, which describe the orientation of the methoxy groups with respect to the attached rings, are given in Table 1.

Weak intramolecular C–H···O and C–H···Br interactions are observed. Intermolecular C–H···O hydrogen bonds help to stabilize the crystal structure (Table 2). In addition, a short Br2···O4(1–x, –y, 1–z) contact of $3.065 (2)$ Å is observed.

Experimental

The title compound was prepared according to the literature procedure of Bringmann *et al.* (2003). The reaction was initiated by the addition of one molar equivalent of (2-bromo-3,4,5-trimethoxyphenyl)methanol, 4-bromo-7-methoxy-1,3-benzodioxole-5-carboxylic acid, dicyclohexylcarbodiimide and 0.2 molar equivalent of 4-(dimethylamino)pyridine to dichloromethane. The mixture was stirred at room temperature for 12 h. A white powder (m.p. 394 K) resulted (yield 88%) and single crystals of (I) suitable for X-ray

analysis were obtained by slow evaporation of a dichloromethane solution.

Crystal data

$C_{19}H_{18}Br_2O_8$ $V = 1009.2(3) \text{ \AA}^3$
 $M_r = 534.15$ $Z = 2$
 Triclinic, $P\bar{1}$ $D_x = 1.758 \text{ Mg m}^{-3}$
 $a = 7.9131(13) \text{ \AA}$ Mo $K\alpha$ radiation $\mu = 4.06 \text{ mm}^{-1}$
 $b = 9.5314(15) \text{ \AA}$ $T = 294(2) \text{ K}$
 $c = 13.949(3) \text{ \AA}$ Block, colourless
 $\alpha = 77.692(12)^\circ$ $0.12 \times 0.10 \times 0.08 \text{ mm}$
 $\beta = 79.132(17)^\circ$
 $\gamma = 88.946(18)^\circ$

Data collection

Rigaku Saturn diffractometer 10738 measured reflections
 ω scans 4595 independent reflections
 Absorption correction: multi-scan 3140 reflections with $I > 2\sigma(I)$
 (Jacobson, 1998) $R_{int} = 0.035$
 $T_{min} = 0.642, T_{max} = 0.737$ $\theta_{max} = 27.5^\circ$

Refinement

Refinement on F^2 $w = 1/[\sigma^2(F_o^2) + (0.0412P)^2]$
 $R[F^2 > 2\sigma(F^2)] = 0.032$ where $P = (F_o^2 + 2F_c^2)/3$
 $wR(F^2) = 0.078$ $(\Delta/\sigma)_{max} = 0.001$
 $S = 0.93$ $\Delta\rho_{max} = 0.62 \text{ e \AA}^{-3}$
 4595 reflections $\Delta\rho_{min} = -0.54 \text{ e \AA}^{-3}$
 264 parameters Extinction correction: *SHELXL97*
 H-atom parameters constrained Extinction coefficient: 0.0077 (10)

Table 1

Selected torsion angles ($^\circ$).

C8—O1—C1—C6	−7.4 (4)	C18—O7—C14—C15	81.7 (3)
C17—O6—C13—C12	86.8 (3)	C19—O8—C15—C16	1.1 (4)

Table 2

Hydrogen-bond geometry ($\text{\AA}, ^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
C6—H6 \cdots O5	0.93	2.37	2.713 (3)	101
C8—H8A \cdots O2 ⁱ	0.96	2.59	3.533 (3)	166
C16—H16 \cdots O5	0.93	2.37	2.735 (3)	103
C17—H17B \cdots O7 ⁱⁱ	0.96	2.57	3.287 (3)	132
C17—H17C \cdots Br2	0.96	2.93	3.468 (3)	117

Symmetry codes: (i) $-x + 2, -y + 2, -z + 1$; (ii) $-x + 1, -y, -z + 2$.

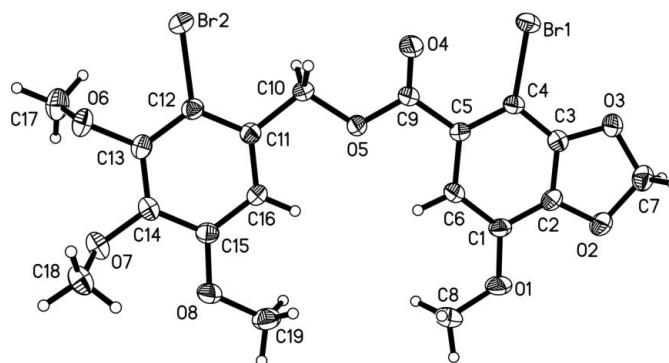


Figure 1

A view of the molecular structure of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 35% probability level and H atoms are represented by circles of arbitrary size.

All H atoms were positioned geometrically and refined as riding, with C—H = 0.93–0.97 \AA . $U_{iso}(H)$ values were set equal to $xU_{eq}(C)$, where $x = 1.5$ for methyl H atoms and 1.2 for all others.

Data collection: *CrystalClear* (Molecular Structure Corporation & Rigaku, 1999); cell refinement: *CrystalClear*; data reduction: *CrystalStructure* (Rigaku/MS, 2005); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *CrystalStructure*.

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